

**TECH OFFER**

## High-Purity Li<sub>2</sub>S Production Method for Next-Generation All-Solid-State Batteries



### KEY INFORMATION

TECHNOLOGY CATEGORY:  
Chemicals - Catalysts

TECHNOLOGY READINESS LEVEL (TRL): **TRL4**  
COUNTRY: **SOUTH KOREA**  
ID NUMBER: **TO175458**

### OVERVIEW

All-solid-state lithium batteries are emerging as the next frontier in energy storage, offering higher safety and energy density than conventional lithium-ion systems. A key challenge in their development lies in producing high-purity lithium sulfide (Li<sub>2</sub>S)—a critical precursor for sulfide solid electrolytes such as Li<sub>10</sub>GeP<sub>2</sub>S<sub>12</sub>. Conventional synthesis methods typically require high temperatures and complex purification, resulting in high costs and limited scalability.

This technology presents a novel low-temperature chemical synthesis process for producing battery-grade Li<sub>2</sub>S under mild reaction conditions (below 100 °C). Using a solution-based approach with organic solvents, surfactants, and catalysts, the process achieves precise control over Li<sub>2</sub>S particle size (50 nm–1 μm) and crystallinity. The resulting material exhibits high purity (up to 99.5% - 99.9%), high yield (85% - 90%) and improved ionic conductivity when incorporated into solid electrolytes. The simplified synthesis eliminates post-annealing and purification steps, reducing production cost and energy use while enabling scalable mass production. There is also no need for dry-room or toxic-gas facility, drastically reducing costs for CAPEX and OPEX.

The technology owner is looking for R&D collaboration with battery manufacturers, material suppliers, and R&D institutions who are developing next-generation all-solid-state batteries.

## TECHNOLOGY FEATURES & SPECIFICATIONS

- **Synthesis method**
  - Low-temperature, solution-based chemical synthesis operating below 100 °C, enabling energy-efficient production of high-purity lithium sulfide (Li<sub>2</sub>S).
- **Process characteristics**
  - Conducted under mild reaction conditions (1–4 hours) in a controlled solvent environment.
  - Achieves uniform particle formation with precise control of particle size (50 nm – 1 μm) and crystallinity.
  - Eliminates post-annealing and complex purification, simplifying downstream processing.
  - Suitable for continuous or batch-type scale-up, compatible with industrial chemical reactors.
- **Product properties:**
  - High-crystallinity, high-purity Li<sub>2</sub>S with minimal oxide or carbonate impurities.
  - Stable morphology supporting homogeneous mixing with sulfide glass or crystalline precursors.
  - Demonstrated high ionic conductivity and enhanced coulombic efficiency when used in solid electrolytes (e.g., Li<sub>10</sub>GeP<sub>2</sub>S<sub>12</sub>-type systems).– Improved capacity retention and cycling stability in all-solid-state lithium cells.
- **Integration:**
  - The Li<sub>2</sub>S material can be readily combined with GeS<sub>2</sub> and P<sub>2</sub>S<sub>5</sub> or other sulfide formers through standard ball-milling and pellet-sintering techniques to fabricate dense, high-performance solid electrolytes.

## POTENTIAL APPLICATIONS

- **All-Solid-State Lithium-Ion Batteries (ASSBs):** Li<sub>2</sub>S serves as a core raw material for sulfide-based solid electrolytes.
- **Advanced Energy Storage Devices:** Applicable to high-energy, safe, nonflammable storage systems for EVs, portable electronics, and grid energy storage.
- **Material Supply Chain Innovation:** Can be integrated into Li<sub>2</sub>S powder manufacturing for solid-state electrolyte production lines.
- **R&D Platforms:** Useful for developing new sulfide-based composite electrolytes and interface-stabilized cathodes.

## UNIQUE VALUE PROPOSITION

Unlike conventional Li<sub>2</sub>S synthesis methods that rely on high-temperature processes (>400 °C) or hazardous gas precursors (H<sub>2</sub>S), this technology employs a low-temperature wet-chemistry approach using readily available, safer precursors.

It offers:

- High crystallinity and purity without the need for annealing
- Controlled particle morphology (flake or spherical) to enhance electrolyte dispersion
- Shorter reaction time (1–4 hours versus >10 hours)
- Simplified, scalable process suitable for mass production and industrial implementation

This combination of process efficiency and material quality results in higher ionic conductivity and greater performance stability in solid-state batteries—delivering a strong competitive advantage for next-generation energy storage manufacturing.